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(1) Applicant: PILKINGTON INSULATION LIMITED Prescot Road St Helens Merseyside WA10 3TT (GB)

18 Claremont Drive
Ormskirk, Lancashire L39 4SP (GB)
Inventor: Mason, Allen Frank
17 Springpool, Near-Wigan
Greater Manchester (GB)
Inventor: Shorrock, Peter
19 Ash Grove, Standish
Wigan, Greater Manchester (GB)
Inventor: Edwards, Norman Anthony
53 The Serpentine, Aughton
Ormskirk, Lancashire L39 6RN (GB)

Representative : Blatchford, William Michael et al Withers & Rogers 4 Dyer's Buildings Holborn London EC1N 2JT (GB)

- (64) Glass composition and use.
- 67 A glass composition capable of being spun into fibres has the following components expressed as weight percent: SiO<sub>2</sub> 66 to 73; Al<sub>2</sub>O<sub>3</sub> 0.85 to 5; R<sub>2</sub>O (= Na<sub>2</sub>O + K<sub>2</sub>O) 14 to 17.5; CaO 6.5 to 12; and SiO<sub>2</sub> + A<sub>2</sub>O<sub>3</sub> 69 to 74. The composition is free of boric oxide and consequently avoids pollution difficulties associated with that compound, yet it can be used in high temperature spinners to produce durable fibres.

This invention relates to a boric oxide free glass composition capable of being spun into fibres, it also relates to a method of spinning compositions according to the invention in a spinner made from a mechanically alloyed or oxide dispersion strengthened alloy and to glass fibre insulation produced from the compositions according to the invention.

Glass compositions are known for use in the technique of fiberizing glass using a centrifugal spinner. The compositions have customarily incorporated boric oxide in order to give them temperature/viscosity characteristics that will enable the glass to pass freely through orifices in the centrifugal spinner wall at a temperature sufficiently low to prevent excessive comosion and erosion of the spinner. A problem with the use of boric oxide is that boron is volatile and may escape from the glass melting tank to cause pollution problems. Furthermore it tends to condense on regenerators thereby fouling them up and preventing use of such devices to improve the thermal efficiency of fuel fired glass melting tanks.

GB 2 041 910 proposes the reduction or elimination of boric oxide with a consequent rise in liquidus temperature and high levels of alumina or baria. These reduced boric oxide compositions are acknowledged to be virtually impossible to fiberize on an industrial basis by the prior art spinning techniques. A technique involving the use of a novel spinner shape is proposed, but, in practice, the corrosion of such a spinner when fabricated from conventional alloy is unacceptably high. Also the glass compositions according to this patent have unacceptably low durability.

US Patent 4 402 767 proposes a novel spinner fabrication process to produce a mechanically alloyed or oxide dispersion strengthened alloy by a combination of warm working, annealing and hot forming processes. Such spinners are claimed to have excellent resistance against molten glass attack and are said to be capable of producing glass fibres and mineral wool at temperatures as high as 1315°C. No details of glass fibre production are given in the specification.

The problem has been to identify boron free glass compositions which have the required durability and can be used in high temperature spinners such as those described in US 4 402 767. Although the absence of boric oxide gives rise to a deterioration in the aqueous durability of the glass, we have identified a range of compositions for which the durability is satisfactory.

According to the present invention a boric oxide free glass composition capable of being spun into fibres, comprises the following components, expressed as weight percent:

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SiO<sub>2</sub> 66-73

30 Al<sub>2</sub>O<sub>3</sub> 0.8-5

R<sub>2</sub>O = Na<sub>2</sub>O + K<sub>2</sub>O 14-17.5

CaO 6.5-12

SiO<sub>2</sub> + Al<sub>2</sub>O<sub>3</sub> 69-74
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Such glasses have a viscosity of 1000 poise or less at temperatures up to about 1200°C and a liquidus at least 130°C below the 1000 poise temperature.

The composition may also contain: 0-2%  $Fe_2O_3$ ; 0-5% MgO; 0-0.6%  $SO_3$ , all expressed as weight percent. Preferably the components are present in weight percentages within the following ranges:

```
SIO<sub>2</sub>
                                 67-72.4
                                 1-4
Al<sub>2</sub>O<sub>2</sub>
                                 14.5-17
R_2O
K<sub>2</sub>O
                                 0.5-2
Na<sub>2</sub>O
                                 13.5-16.5
CaO
                                 7-11.2
                                 0.1 - 2.5
Fe<sub>2</sub>O<sub>3</sub>
MgO
                                 0.2 - 4.4
SiO<sub>2</sub> + Al<sub>2</sub>O<sub>3</sub>
                                 70-73.7
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Most preferably the components are present in weight percentages within the following ranges

```
SIO<sub>2</sub>
                                    67-70
Al<sub>2</sub>O<sub>3</sub>
                                    2-4
Na<sub>2</sub>O
                                    14-15.5
                                    0.5 - 1.5
K<sub>2</sub>O
MgO
                                    3-4.5
                                    7-8.5
CaO
Fe<sub>2</sub>O<sub>3</sub>
                                    0.3-2
                                   0-0.3
SO<sub>3</sub>
SiO<sub>2</sub> + Al<sub>2</sub>O<sub>3</sub>
                                   70-72
Na<sub>2</sub>O + K<sub>2</sub>O
                                    15-16.5
```

A high level of Na<sub>2</sub>O is needed to give a low liquidus temperature but the preferred narrow range of MgO

allows the Na<sub>2</sub>O to be kept to a minimum so optimising durability while maintaining liquidus at least 160 centigrade degrees below the 1000 poise viscosity temperature, which is advantageously below 1170°C.

The invention will now be described with reference to the following non-limiting examples 1-28. Low levels of  $SO_3$ ,  $K_2O$  and  $Fe_2O_3$  are recorded as a minimum in these examples. In fact they are only present at the lower levels as impurities in the raw materials and do not make a significant contribution to the properties of the glass fibres. Details of the examples and durability tests are given in the Table. It should be noted that Example 18 falls outside the scope of the present invention as its low level of alumina renders it insufficiently durable.

There is no established international test procedure for assessing the suitability of a glass for glass fibre insulation applications. It has become the practice to use the laboratory ware test of ISO 719 as it gives a useful guide to the aqueous durability and weathering resistance of the glass. In this test 2g of glass grains are treated with distilled water for 60 mins at 98°C and the extracted alkali titrated against 0.01 M HCL. The durability is described in terms of the alkali extracted per gram of glass as calculated from the acid required to neutralise. The relevant classes for glass wool are ISO Class 3 - from 62 to 264 micrograms of alkali per gram; and Class 4 - from 265 to 620 micrograms per gram. Glass wool is preferably in Class 3 but a good Class 4 may be acceptable because the performance of the insulation is a function of both glass and resin binder, and binders are available for use with higher release glasses. A class 3 rating is good and a class 4 rating is acceptable, but not as good.

Table 1

	74	<b>ग</b>	67	60	2	n	6	4	S	-	2	2	~	8	S	4
			9		7	0.3			15.5		16.5	0.2	1142	1018	285	
	6		67	3	70	0.3	8	4	16.5	1	17.5	0.2	1136	086	406	7
	4		69	3	72	0.3	8	4	14.5	1	15.5	0.2	1173	1002	251	3
	<u> </u>	7	20	3	73	0.3	8	4	13.5	1	14.5	0.2	1191	1012	186	67
	-	7	29	4	71	0.3	В	4	15.5	7	16.5	0.2	1159	1002	248	•
_	6	2	89	က	71	0.3	80	9	15.5	1	16.5	0.2	1155	982	260	6
	C	7	69	7	71	0.3	8	4	15.5	1	16.5	0.2	1150	964	270	P
		4	70	1	71	0.3	8	4	15.5	П	16.5	0.2	1146	952	307	V
		ехащоте	S102	A1203	S102+A1203	Fe203	CaO	MgO	Na20	K20	R20	803	Tlog3	liquidus	ugR20/g	0001

10 11	12	13	14	15	91
		72.4		89	68
	E	1.3		4	3
	71	73.7		72	71
	1.3	0.1	6.0		1.9
8 7.5		11.2			
4		0.2		3.5	3.9
	15	14		15.5	14
	-	0.5		1	1
	16	14.5	:	16.5	15
		0.2			0.2
		1170			1170
		1018			1026
	253	286			195
	4	4	9	8	. 3
68.5 0.3 1 1 1 1 1 1 1 1 1 1 1 2 3 3	15 15 16 2 2 2 5 9 6 6 9 3 3 3 3 3 3 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 6 6 9 9 9 6 6 9 9 9 6 6 9	69 68 3 72 71 0.3 11.3 7.5 7.5 4 4 4 15 15 15 16 16 0.2 0.2 1169 1163 966 1000 966 1000 3 4 4	68 3 3 1.3 7.5 7.5 1.3 1.6 1.6 1.000 1.000 2.53	68 72.4 3 1.3 71 73.7 1.3 0.1 7.5 11.2 4 0.2 15 14 1 0.5 16 14.5 16 14.5 16 14.5 16 14.5 16 14.5 16 2 0.2 16 2 0.2 16 2 0.2 16 3 3 3 3 4 4	68 72.4 70 3 1.3 1.3 3 7.5 11.2 0.1 0.3 7.5 11.2 8.3 7.5 11.2 8.3 1.5 11.2 8.3 1.0 0.2 3.9 1.0 0.2 0.9 1.0 0.2 0.9 1.0 0.2 0.2 1.0 0.2 0.2

ace 2

10 15 20 25 1 elqeL 35		
20 25 30 35	10	
25 30 35 40	15	
35 35 40	20	
<b>35</b> <b>4</b> 0	26	
40	30	Table 1
	35	
45	40	
	45	

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Example	17	18	19	20	21	22	23	24
						-		
S102	68.5	7.1	89	69	89	67	99	99
A1203	3	0	က	3	3	က	3	4
S102+A1203	71.5	71	7.1	72	71	20	69	70
Fe203		0.3	0.3	0.3	1.3	2.3	3.3	2.5
CaO	7.6	8	8	7.5	7.5	7.5	7.5	7.8
Mao	4.4	4	4	4	4	4	4	Э
Na20	15	15.5	15.5	15	15	15	15	15.5
K20	1	-	1	П	1	F-1	1	1
R20	16	16.5	16.5	16	91	16	16	16.5
503	0.2	0.2	0.2	0.2	0.2	-	0.2	0.2
Tlog3	1165	1142		1170	1164		1150	1155
liguidus	1006	940		974	966	1002	1024	1010
ugR20/a	251	474	251	259	253	236	239	240
Class	9	P	3	ဇ	3	က	3	က

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1.5 1.5 10.2 1014 206

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Table

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Claims

A boric oxide free glass composition capable of being spun into fibres, comprising the following components expressed as weight percent:

SIO2

68-73

Sxample

Al<sub>2</sub>O<sub>3</sub>

0.8-5

 $R_2O = Na_2O + K_2O$ CaO 14-17.5 6.5-12

45 SIO<sub>2</sub> + Al<sub>2</sub>O<sub>3</sub>

69-74

A glass composition according to claim 1, further comprising the following components expressed as weight percent:

Fe<sub>2</sub>O<sub>3</sub> 0-2

MgO

O 0-5

SO<sub>3</sub> 0-0.6

3. A glass composition according to claim 1 or 2, wherein the components are present in weight percentages within the following ranges:

SiO<sub>2</sub>

67-72.4

Al<sub>2</sub>O<sub>3</sub>

1-4

R<sub>2</sub>O K<sub>2</sub>O 14.5-17 0.5-2

	Na <sub>2</sub> O	13.5-16.5
	CaO	7-11.2
	Fe <sub>2</sub> O <sub>3</sub>	0.1-2.5
	MgO	0.2-4.4
5	SiO <sub>2</sub> + Al <sub>2</sub> O <sub>3</sub>	70-73.7

4. A glass composition according to claim 3, wherein the components are present in weight percentages within the following ranges:

	SiO <sub>2</sub>	67-70
10	Al <sub>2</sub> O <sub>3</sub>	2-4
	Na <sub>2</sub> O	14-15.5
	K₂O	0.5-1.5
	MgO	3-4.5
	CaO	7-8.5
15	Fe <sub>2</sub> O <sub>3</sub>	0.3-2
	SO <sub>3</sub>	0-0.3
	SIO <sub>2</sub> + Al <sub>2</sub> O <sub>3</sub>	70-72
	Na <sub>2</sub> O + K <sub>2</sub> O	15-16.5

- 5. An insulating glass fibre production process including the spinning of molten glass at a high temperature in a spinner made from a mechanically alloyed or oxide dispersion strengthened alloy to produce fibres having a composition according to any preceding claim.
  - 6. Insulating glass fibres having a composition according to any one of claims 1 to 4.

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# EUROPEAN SEARCH REPORT

Application Number

EP 92 30 4660

	Citation of decree	IDERED TO BE RELEVA	111	
Category	of recevant p		Rejevant to claim	CLASSIFICATION OF THE APPLICATION (Int. CL5)
X	EP-A-0 399 320 (BAYER * abstract *	AG,)	1-4,5	C03C13/00
۲	_		5	C03C3/087
x, a	US-A-4 203 746 (J.A. B	ATTIGELLI ET AL.)	1-4,6	
۲			5	
×	CHENICAL ABSTRACTS, vo 1982, Columbus, Ohio, abstract no. 167698Y, M. CZAJA ET AL.: 'Rest technological and desi production of glass fi	US; ilts of the selection of gn parameters for	1-4,6	·
۲	page 299 ; * abstract *		5	
D,Y	US-A-4 402 767 (J.W. H	 INZE ET AL.)	5	
		****		TECHNICAL FIELDS SEARCHED (Int. Cl.5 )
				CO3C
	The present search report has b			
•	THE HAGUE	Data of completion of the search OB SEPTEMBER 1992	REED	Domine TJK A, M, E,
X : parti Y : parti docu A : tech	ATEGORY OF CITED DOCUME cularly relevant if taken alone cularly relevant if combined with an ment of the tame category nological background	E : earlier palem after the filh other D : document of L : document of	nciple underlying the document, but public g date at in the application of for other reasons	invention shed on, or
C: non-	written disclosure mediate document	A: member of the document	ie same patent family	, corresponding

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